impact on a substantial number of small entities under the Regulatory Flexibility Act (5 U.S.C. 601 et seq.). Because this rule approves pre-existing requirements under State law and does not impose any additional enforceable duty beyond that required by State law, it does not contain any unfunded mandate or significantly or uniquely affect small governments, as described in the Unfunded Mandates Reform Act of 1995 (Pub. L. 104-4). This rule also does not have Tribal implications because it will not have a substantial direct effect on one or more Indian Tribes, on the relationship between the Federal Government and Indian Tribes, or on the distribution of power and responsibilities between the Federal Government and Indian Tribes, as specified by Executive Order 13175 (65 FR 67249, November 9, 2000). This action also does not have Federalism implications because it does not have substantial direct effects on the States, on the relationship between the national government and the States, or on the distribution of power and responsibilities among the various levels of government, as specified in Executive Order 13132 (64 FR 43255, August 10, 1999). This action merely approves a State rule implementing a Federal requirement, and does not alter the relationship or the distribution of power and responsibilities established in the CAA. This rule also is not subject to Executive Order 13045 "Protection of Children from Environmental Health Risks and Safety Risks" (62 FR 19885, April 23, 1997), because it approves a State rule implementing a Federal standard. In reviewing section 111(d)/ 129 plan submissions, EPA's role is to approve State choices, provided that they meet the criteria of the CAA. In this context, in the absence of a prior existing requirement for the State to use voluntary consensus standards (VCS), EPA has no authority to disapprove a 111(d)/129 plan submission for failure to use VCS. It would thus be inconsistent with applicable law for EPA, when it reviews a 111(d)/129 plan submission, to use VCS in place of a 111(d)/129 plan submission that otherwise satisfies the provisions of the CAA. Thus, the requirements of section 12(d) of the National Technology Transfer and Advancement Act of 1995 (15 U.S.C. 272 note) do not apply. This rule does not impose an information collection burden under the provisions of the Paperwork Reduction Act of 1995 (44 U.S.C. 3501 et seq.).

B. Submission to Congress and the Comptroller General

The Congressional Review Act, 5 U.S.C. 801 et seq., as added by the Small **Business Regulatory Enforcement** Fairness Act of 1996, generally provides that before a rule may take effect, the agency promulgating the rule must submit a rule report, which includes a copy of the rule, to each House of the Congress and to the Comptroller General of the United States. EPA will submit a report containing this rule and other required information to the U.S. Senate, the U.S. House of Representatives, and the Comptroller General of the United States prior to publication of the rule in the Federal Register. This rule is not a "major rule" as defined by 5 U.S.C. 804(2).

C. Petitions for Judicial Review

Under section 307(b)(1) of the Clean Air Act, petitions for judicial review of this action must be filed in the United States Court of Appeals for the appropriate circuit by October 2, 2009. Filing a petition for reconsideration by the Administrator of this final rule does not affect the finality of this rule for the purposes of judicial review nor does it extend the time within which a petition for judicial review may be filed, and shall not postpone the effectiveness of such rule or action. Parties with objections to this direct final rule are encouraged to file a comment in response to the parallel notice of proposed rulemaking for this action published in the proposed rules section of today's Federal Register, rather than file an immediate petition for judicial review of this direct final rule, so that EPA can withdraw this direct final rule and address the comment in the proposed rulemaking. This action, approving the submitted West Virginia HMIWI plan revision, may not be challenged later in proceedings to enforce its requirements. (See section 307(b)(2).)

List of Subjects in 40 CFR Part 62

Environmental protection, Administrative practice and procedure, Air pollution control, Aluminum, Fertilizers, Fluoride, Intergovernmental relations, Paper and paper products industry, Phosphate, Reporting and recordkeeping requirements, Sulfur oxides, Sulfur acid plants, Waste treatment and disposal.

Dated: July 21, 2009.

William C. Early,

Acting Regional Administrator, Region III.

■ 40 CFR Part 62, Subpart XX, is amended as follows:

PART 62-[AMENDED]

■ 1. The authority citation for part 62 continues to read as follows:

Authority: 42 U.S.C. 7401 et seq.

Subpart XX—West Virginia

■ 2. Section 62.12150 is amended by designating the existing paragraph as paragraph (a) and adding paragraph (b) to read as follows:

§ 62.12150 Identification of plan. *

* *

*

(b) On May 11, 2009, the West Virginia Department of Environmental Protection submitted a State plan revision (#1) that consolidates all existing section 111(d)/129 incinerator regulatory requirements into one modified rule, WV45CSR18.

■ 3. Section 62.12152 is amended by designating the existing paragraph as paragraph (a) and adding paragraph (b) to read as follows:

§61.12152 Effective date. *

*

* (b) Plan revision #1 is effective October 2, 2009.

[FR Doc. E9–18482 Filed 7–31–09; 8:45 am] BILLING CODE 6560-50-P

ENVIRONMENTAL PROTECTION AGENCY

40 CFR Part 141

[EPA-HQ-OW-2009-0345; FRL-8930-8]

Expedited Approval of Alternative Test Procedures for the Analysis of **Contaminants Under the Safe Drinking** Water Act; Analysis and Sampling **Procedures**

AGENCY: Environmental Protection Agency (EPA). **ACTION:** Final rule.

SUMMARY: This action announces the **Environmental Protection Agency's** (EPA's) approval of alternative testing methods for use in measuring the levels of contaminants in drinking water and determining compliance with national primary drinking water regulations. The Safe Drinking Water Act (SDWA) authorizes EPA to approve the use of alternative testing methods through publication in the Federal Register. EPA is using this streamlined authority to make six additional methods available for analyzing drinking water samples required by regulation. This expedited approach provides public water systems, laboratories, and primacy agencies with more timely access to new measurement techniques and greater flexibility in the selection of analytical methods, thereby reducing monitoring costs while maintaining public health protection.

DATES: This action is effective August 3, 2009.

FOR FURTHER INFORMATION CONTACT: Safe Drinking Water Hotline (800) 426-4791 or Patricia Snyder Fair, Technical Support Center, Office of Ground Water and Drinking Water (MS 140), Environmental Protection Agency, 26 West Martin Luther King Drive, Cincinnati, OH 45268; telephone number: (513) 569–7937; e-mail address: fair.pat@epa.gov.

SUPPLEMENTARY INFORMATION:

I. General Information

A. Does This Action Apply to Me?

Public water systems are the regulated entities required to measure contaminants in drinking water samples. In addition, EPA Regions as well as States and Tribal governments with authority to administer the regulatory program for public water systems under SDWA may also measure contaminants in water samples. When EPA sets a monitoring requirement in its national primary drinking water regulations for a given contaminant, the Agency also establishes in the

regulations standardized test procedures for analysis of the contaminant. This action makes alternative testing methods available for particular drinking water contaminants beyond the testing methods currently established in the regulations. EPA is providing public water systems required to test water samples with a choice of using either a test procedure already established in the existing regulations or an alternative test procedure that has been approved in this action. Categories and entities that may ultimately be affected by this action include:

Category	Examples of potentially regulated entities	NAICS ¹
State, Local, & Tribal Governments	States, local and tribal governments that analyze water samples on behalf of public water systems required to conduct such analysis; States, local and tribal governments that themselves operate community and non-transient non-community water systems required to monitor.	924110
Industry	Private operators of community and non-transient non-community water systems re- guired to monitor.	221310
Municipalities	Municipal operators of community and non-transient non-community water systems re- quired to monitor.	924110

¹ North American Industry Classification System.

This table is not exhaustive, but rather provides a guide for readers regarding entities likely to be affected by this action. This table lists the types of entities that EPA is now aware could potentially be affected by this action. Other types of entities not listed in the table could also be impacted. To determine whether your facility is affected by this action, you should carefully examine the applicability language at 40 CFR 141.2 (definition of public water system). If you have questions regarding the applicability of this action to a particular entity, consult the person listed in the preceding FOR FURTHER INFORMATION CONTACT section.

B. How Can I Get Copies of This Document and Other Related Information?

1. Docket. EPA established a docket for this action under Docket ID No. EPA-HQ-OW-2009-0345. Publicly available docket materials are available either electronically through www.regulations.gov or in hard copy at the Water Docket in the EPA Docket Center, (EPA/DC) EPA West, Room 3334, 1301 Constitution Ave., NW., Washington, DC. Copyrighted materials are available only in hard copy. The EPA Docket Center Public Reading Room is open from 8:30 a.m. to 4:30 p.m., Monday through Friday, excluding legal holidays. The telephone number for the Public Reading Room is (202)

566–1744, and the telephone number for the Water Docket is (202) 566-2426.

2. Electronic Access. You may access this Federal Register document electronically through the EPA Internet under the "Federal Register" listings at http://www.epa.gov/fedrgstr/.

Abbreviations and Acronyms Used in This Action

- CFR: Code of Federal Regulations
- DBCP: Dibromochloropropane
- EDB: Ethylene Dibromide

EPA: Environmental Protection Agency

GC: Gas Chromatography

- LED: Light-Emitting Diode
- MS: Mass Spectrometry

NEMI: National Environmental Methods

Index nm: Nanometers

SDWA: Safe Drinking Water Act

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II. Background

A. What Is the Purpose of This Action?

In this action, EPA is approving six analytical methods for determining

contaminant concentrations in samples collected under SDWA. Regulated parties required to sample and monitor may use either the testing methods already established in existing regulations or the alternative testing methods being approved in this action. The new methods are listed in Appendix A to Subpart C in 40 CFR 141 and on EPA's drinking water methods Web site at http://www.epa.gov/ safewater/methods/ analyticalmethods expedited.html.

B. What Is the Basis for This Action?

When EPA determines an alternative analytical method is "equally effective" (i.e., as effective as a method that has already been promulgated in the regulations), SDWA allows EPA to approve the use of the alternative method through publication in the Federal Register. See Section 1401(1) of SDWA. EPA is using this streamlined approval authority to make six additional methods available for determining contaminant concentrations in samples collected under SDWA. EPA has determined that, for each contaminant or group of contaminants listed in Section III, the additional testing methods being approved in this action are equally as effective as one or more of the testing methods already established in the regulations for those contaminants. Section 1401(1) states that the newly approved methods "shall be treated as

an alternative for public water systems to the quality control and testing procedures listed in the regulation." Accordingly, this action makes these additional (and optional) six analytical methods legally available for meeting EPA's monitoring requirements.

This action does not add regulatory language, but does, for informational purposes, update an appendix to the regulations at 40 CFR part 141 that lists all methods approved under Section 1401(1) of SDWA. Accordingly, while this action is not a rule, it is updating CFR text and therefore is being published in the "Final Rules" section of this **Federal Register**.

EPA described this expedited methods approval process in an April 10, 2007, **Federal Register** notice (72 FR 17902) (USEPA 2007) and announced its intent to begin using the process. EPA published the first set of approvals in a June 3, 2008, **Federal Register** notice (73 FR 31616) (USEPA 2008) and added Appendix A to 40 CFR Part 141, Subpart C. This action adds six additional methods to Appendix A to Subpart C.

III. Summary of Approvals

EPA is approving six methods that are equally effective relative to methods previously promulgated in the regulations. By means of this notice, these six methods are added to Appendix A of 40 CFR Part 141, Subpart C. For convenience of the reader, the revised Appendix A in its entirety is shown below. However, the only change made to Appendix A through this action is the inclusion of these six additional methods as described in this preamble.

A. Methods Developed by EPA

EPA Method 524.3, Version 1.0. This is a gas chromatography/mass spectrometry (GC/MS) method for the determination of purgeable organic compounds in finished drinking waters. The method analytes are purged from the water sample using helium and trapped on a sorbent material. After purging, the trap is heated and back flushed with helium to transfer the analytes to a capillary GC column. Compounds eluting from the GC are directed into a mass spectrometer for mass analysis and detection. The analytes are identified by comparing the acquired mass spectra and retention times to reference spectra and retention times for calibration standards acquired under identical GC/MS conditions. The concentration of each target analyte is calculated using the internal standard technique and response curves obtained via procedural calibration. The expansion of the method to include the

option of selective ion monitoring makes this method sufficiently sensitive to measure dibromochloropropane (DBCP) and ethylene dibromide (EDB) at the concentrations required for drinking water compliance monitoring.

EPA Method 524.3 is an updated version of EPA Method 524.2, Revision 4.1 (USEPA 1995a), which is currently approved for analyses of compliance samples for 21 volatile organic contaminants and total trihalomethanes. The method development work is described in the method research summary (Zaffiro et al. 2009). The advantages of the new method include:

• Use of maleic acid, a common food preservative, to preserve samples, eliminating the requirement to ship a hazardous reagent (hydrochloric acid) to the field;

• Incorporation of features that allow users to take advantage of modern instrumentation to improve speed and data quality;

• Increased flexibility in selection of method operating parameters; and

• Addition of Method 524.3 as an approved method for DBCP and EDB.

Approved methods for volatile organic contaminants and total trihalomethanes are listed at 40 CFR 141.24(e). EPA Methods 502.2; Revision 2.1 (USEPA 1995b) and 524.2; Revision 4.1 (USEPA 1995a) are approved for benzene; carbon tetrachloride; chlorobenzene; 1,2-dichlorobenzene; 1.4-dichlorobenzene: 1.2dichloroethane; cis-dichloroethylene; trans-dichloroethylene; dichloromethane; 1,2-dichloropropane; ethylbenzene; styrene; tetrachloroethylene; 1,1,1trichloroethane; trichloroethylene; toluene; 1,2,4-trichlorobenzene; 1,1dichloroethylene; 1,1,2-trichlorethane; vinyl chloride; xylenes (total-measured as sum of o-xylene; m-xylene and pxvlene); and total trihalomethanes (sum of chloroform; bromodichloromethane; dibromochloromethane; and bromoform). EPA Method 551.1 (USEPA 1995c) is approved for carbon tetrachloride; tetrachloroethylene; 1,1,1trichloroethane; trichloroethylene; EDB; DBCP; and total trihalomethanes. EPA Method 504.1, Revision 1.1 (USEPA 1995d) is approved for EDB and DBCP. Approved methods for total trihalomethanes are also listed at 40 CFR 141.131(b)(1). For each of the 24 contaminants, the performance characteristics of EPA Method 524.3 were compared to the characteristics of each of the methods currently listed in the regulations as approved for that contaminant (Munch 2009). EPA has determined that, for each of the 24 contaminants, EPA Method 524.3 is

equally as effective for measuring the contaminant as the methods currently listed in the regulations as approved for that contaminant. The basis for this determination is discussed in Munch 2009. EPA is therefore approving use of Method 524.3 for the above named 24 contaminants when analyzing drinking water compliance samples.

EPA Method 524.3 Version 1.0 (USEPA 2009) can be accessed and downloaded directly on-line at http:// epa.gov/safewater/methods/ analyticalmethods ogwdw.html.

B. Methods Developed by Vendors

1. Mitchell Method M5271. Mitchell Method M5271 (Mitchell 2009a) uses laser nephelometry to measure turbidity in drinking water. The method is based on a comparison of the intensity of light scattered by the sample under defined conditions with the intensity of light scattered by a standard reference suspension. Readings are made using an on-line laser nephelometer with the following design criteria:

• Laser light source is monochromatic operated at a nominal wavelength of 650 ± 30nm;

• Incident radiation and any convergence does not exceed ± 1.5 degrees in the measurement area;

• Distance traversed by incident light and scattered light does not exceed 10cm;

• Detector/light receiver is centered at 90 ± 1.5 degrees to the incident light path and the light cone does not exceed ± 30 degrees from 90 degrees; and

• Instrument incorporates a bubble trap and anti-fog windows. Sensor is horizontal and the windows are vertical. Windows are immersed in the sample stream.

Four approved methods for turbidity are listed at 40 CFR 141.74(a)(1). The performance characteristics of Mitchell Method M5271 were compared to the performance characteristics of approved EPA Method 180.1 (USEPA 1993a). The validation study report (Mitchell 2008a) summarizes the results obtained from the turbidimeters placed in series at three different public water systems. One water system used ground water and the other two plants used surface water sources. Measurements included at least one filter backwash at each of the surface water plants.

EPA has determined that the Mitchell Method M5271 is equally effective relative to EPA Method 180.1 that is already promulgated in the regulations at 40 CFR 141.74(a)(1). The basis for this determination is discussed in Wendelken 2009a. Therefore, EPA is approving the Mitchell Method M5271 for determining turbidity in drinking water. A copy of the method can be downloaded from the National Environmental Methods Index (NEMI) at *http://www.nemi.gov* or obtained by contacting Leck Mitchell, PhD, PE, 656 Independence Valley Dr., Grand Junction, CO 81507.

2. Mitchell Method M5331. Mitchell Method M5331 (Mitchell 2009b) uses light-emitting diode (LED) nephelometry to measure turbidity in drinking water. The method is based on a comparison of the intensity of light scattered by the sample under defined conditions with the intensity of light scattered by a standard reference suspension. Readings are made using an on-line LED nephelometer with the following design criteria:

• LED light source is monochromatic operated at a nominal wavelength of 525 ± 15nm;

• Incident radiation and any convergence does not exceed ± 1.5 degrees in the measurement area;

• Distance traversed by incident light and scattered light does not exceed 10cm:

• Detector/light receiver is centered at 90 \pm 1.5 degrees to the incident light path and the light cone does not exceed \pm 30 degrees from 90 degrees; and

• Instrument incorporates a bubble trap and anti-fog windows. Sensor is horizontal and the windows are vertical. Windows are immersed in the sample stream.

Four approved methods for turbidity are listed at 40 CFR 141.74(a)(1). The performance characteristics of Mitchell Method M5331 were compared to the performance characteristics of approved EPA Method 180.1 (USEPA 1993a). The validation study report (Mitchell 2008b) summarizes the results obtained from the turbidimeters placed in series at three different public water systems. One water system used ground water and the other two plants used surface water sources. Measurements included at least one filter backwash at each of the surface water plants.

EPA has determined that the Mitchell Method M5331 is equally effective relative to EPA Method 180.1 that is already promulgated in the regulations at 40 CFR 141.74(a)(1). The basis for this determination is discussed in Wendelken 2009b. Therefore, EPA is approving it for determining turbidity in drinking water. A copy of the method can be downloaded from NEMI at *http://www.nemi.gov* or obtained from Leck Mitchell, PhD, PE, 656 Independence Valley Dr., Grand Junction, CO 81507.

3. Orion Method AQ4500. Thermo Scientific's Orion Method AQ4500 (Thermo Scientific 2009) uses LED nephelometry to measure turbidity in drinking water. The method is based on a comparison of the intensity of light scattered by the sample at 90 degrees to the beam path with the intensity of light scattered by a standard reference suspension. Readings are made using a portable LED nephelometer with the following design criteria:

• White LED light source emits broadband light having peak intensities in the 400nm to 600nm range;

• Distance traversed by incident light and scattered light does not exceed 10cm;

• Detector/light receiver is centered at 90 degrees to the incident light path and the light cone does not exceed \pm 30 degrees from 90 degrees. The detector has spectral peak response between 400nm and 600nm;

• Pulsed light allows for synchronous detection, a technique by which ambient stray light leakage, as well as other electronic induced errors, are effectively cancelled out; and

• Color compensation is achieved using a dual-beam system with two photo detectors.

Four approved methods for turbidity are listed at 40 CFR 141.74(a)(1). The performance characteristics of Thermo Scientific's Orion Method AQ4500 were compared to the performance characteristics of EPA Method 180.1 (USEPA 1993a) listed at 40 CFR 141.74(a)(1) for measurement of turbidity. Two rounds of testing were conducted (Wendelken 2009c). The first was an ASTM round robin study comparing results from analyses of 28 samples of various types using turbidimeters with tungsten filament light sources as specified in EPA Method 180.1 and white LEDs as specified in Thermo Scientific Orion Method AQ4500. A second study involved demonstration of performance at turbidities below 2 nephelometric turbidity units.

EPA has determined that Thermo Scientific's Orion Method AQ4500 is equally effective relative to EPA Method 180.1, which is already promulgated in the regulations at 40 CFR 141.7 $\overline{4}(a)(1)$. The basis for this determination is discussed in Wendelken 2009c. Therefore, EPA is approving Method AQ4500 for the measurement of turbidity in drinking water. A copy of the method can be downloaded from NEMI at http://www.nemi.gov or obtained from Thermo Scientific, 166 Cummings Center, Beverly, MA 01915, Phone: (800) 225-1480, www.thermo.com.

4. Systea Easy (1-Reagent). Systea Scientific, LLC's Systea Easy (1-Reagent) Nitrate Method uses automated discreet analysis by spectrophotometry to determine concentrations of nitrate and nitrite combined or individually in drinking water. The method involves the following steps:

• Reduction of nitrate in a sample to nitrite using a non-hazardous proprietary reagent;

• Diazotizing the nitrite originally in the sample plus the reduced nitrate with sulfanilamide followed by coupling with N-(1-napthyl)ethylenediamine dihydrochloride under acidic conditions to form a highly colored azo dye;

• Colorimetric determination in which the absorbance of color at 546nm is directly proportional to the concentration of the nitrite plus the reduced nitrate in the sample;

• Measurement of nitrite individually by analysis of the sample while eliminating the reduction step; and

• Subtraction of the nitrite value from that of the combined nitrate plus nitrite value to determine nitrate individually.

Approved methods for nitrate and nitrite are listed at 40 CFR 141.23(k)(1). An inter-laboratory study (Systea Scientific, LLC. 2008) was conducted to compare the performance characteristics of the Systea Easy (1-Reagent) Nitrate Method to the characteristics of the EPA Method 353.2 (USEPA 1993b) and Standard Method 4500-NO₃ - F-00 (APHA 1997), which are listed at 40 CFR 141.23(k)(1) for nitrate and nitrite. Ten laboratories analyzed a variety of sample matrices using approved methods. The samples were also analyzed using the Systea Easy (1-Reagent) Nitrate Method.

EPA has determined that the Systea Easy (1-Reagent) Nitrate Method is equally effective relative to EPA Method 353.2 and Standard Method 4500–NO₃⁻ F–00, which are already promulgated in the regulations. The basis for this determination is discussed in Wendelken 2009d. The method is a "green" alternative to other approved methods, which use cadmium, a known carcinogen, for the reduction of nitrate to nitrite. EPA is approving this method for determining nitrate and nitrite concentrations in drinking water to comply with 40 CFR 141.23.

Systea Easy (1-Reagent) Nitrate Method (Systea Scientific, LLC. 2009) can be downloaded from NEMI at *http://www.nemi.gov* or obtained from Systea Scientific, LLC, 900 Jorie Blvd., Suite 35, Oak Brook, IL 60523, Phone: (630) 645–0600.

5. Method ME355.01. "Determination of Cyanide in Drinking Water by GC/MS Headspace" (Eaton 2009) uses direct headspace injection after acidification followed by Gas Chromatography/Mass Spectrometry (GC/MS) to determine the concentration of cyanide, as free cyanide, in drinking water. The method involves the following steps:

• Acidification of the sample;

• Heating the sample to 60 degrees Celsius with agitation;

• Direct injection of 1 milliliter of headspace onto the nitrogen cooled cryotrap; and

• Analysis using temperature programmed GC/MS.

The performance characteristics of Method ME355.01 were determined in three laboratories by replicate analyses of fortified samples (Wendelken 2009e). The results were compared to the characteristics of EPA Method 335.4 (USEPA 1993c) and Standard Method 4500–CN⁻ F–99 (APHA 1999) listed at 40 CFR 141.23(k)(1) for cyanide. EPA has determined that Method ME355.01 is equally effective relative to each of these two methods. The basis for this determination is discussed in Wendelken 2009e. Therefore, EPA is approving this method for determining cyanide concentrations in drinking water to comply with 40 CFR 141.23.

Method MÉ335.01 can be downloaded from NEMI at *http://www.nemi.gov* or obtained from James Eaton, PhD, H & E Testing Laboratory, 221 State Street, Augusta, ME 04333, Phone: (207) 187– 2727.

IV. Statutory and Executive Order Reviews

As noted above, under the terms of SDWA Section 1401(1), this streamlined method approval action is not a rule. Accordingly, the Congressional Review Act, 5 U.S.C. 801 et seq., as added by the Small Business Regulatory Enforcement Fairness Act of 1996, does not apply because this action is not a rule for purposes of 5 U.S.C. 804(3). Similarly, this action is not subject to the Regulatory Flexibility Act because it is not subject to notice and comment requirements under the Administrative Procedure Act or any other statute. In addition, because this approval action is not a rule but simply makes alternative (optional) testing methods available for monitoring under SDWA, EPA has concluded that other statutes and executive orders generally applicable to rulemaking do not apply to this approval action.

V. References

American Public Health Association (APHA), 2000. Standard Method 4500– NO₃ [–] F–00. Automated Cadmium Reduction Method. Approved by Standard Methods Committee 2000. Standard Methods Online. (Available at http://www.standardmethods.org.) American Public Health Association (APHA), 1999. Standard Method 4500– CN⁻ F–99. Cyanide-Selective Electrode Method. Approved by Standard Methods Committee 1999. Standard Methods Online. (Available at *http:// www.standardmethods.org.*)

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Munch, D., 2009. Memo to the record describing basis for expedited approval of EPA Method 524.3. May 22, 2009.

Systea Scientific, LLC., 2008. Validation Study Report for New Method Approval of Nitrate Analysis in Wastewater and Drinking Water Utilizing Systea Scientific, LLC Non-Hazardous Proprietary Reagent R1, Systea Easy (1-Reagent) Nitrate 0.050–10 mg/L. September 15, 2008. 900 Jorie Blvd., Suite 35, Oak Brook, IL 60523.

Systea Scientific, LLC., 2009. Systea Easy (1-Reagent) Nitrate Method, February 4, 2009. 900 Jorie Blvd., Suite 35, Oak Brook, IL 60523. (Available at *http://www.nemi.gov.*)

Thermo Scientific, 2009. Orion Method AQ4500, Revision 1.0. Determination of Turbidity by LED Nephelometry, May 8, 2009. 166 Cummings Center, Beverly, MA 01915. (Available at http://www.nemi.gov.)

USEPA. 1993a. EPA Method 180.1, Revision 2.0, "Determination of Turbidity by Nephelometry" in Methods for the Determination of Inorganic Substances in Environmental Samples, EPA/600/R–93/100. (Available at http://www.nemi.gov.)

USEPA. 1993b. EPA Method 353.2, Revision 2.0, "Determination of Nitrate-Nitrite Nitrogen by Automated Colorimetry" in Methods for the Determination of Inorganic Substances in Environmental Samples, EPA/600/R– 93/100. (Available at *http:// www.nemi.gov.*)

USEPA. 1993c. EPA Method 335.4, Revision 1.0, "Determination of Total Cyanide by Semi-Automated Colorimetry" in Methods for the Determination of Inorganic Substances in Environmental Samples, EPA/600/R– 93/100. (Available at *http:// www.nemi.gov.*)

USEPA. 1995a. EPA Method 524.2, Revision 4.1, "Measurement of Purgeable Organic Compounds in Water by Capillary Column Gas Chromatography/Mass Spectrometry" in Methods for the Determination of Organic Compounds in Drinking Water—Supplement III, EPA/600/R–95– 131. (Available at http://www.nemi.gov.)

USEPA. 1995b. EPA Method 502.2, Revision 2.1, "Volatile Organic Compounds in Water by Purge and Trap Capillary Column Gas Chromatography with Photoionization and Electrolytic Conductivity Detectors in Series" in Methods for the Determination of Organic Compounds in Drinking Water—Supplement III, EPA/600/R–95– 131. (Available at http://www.nemi.gov.)

USEPA. 1995c. EPA Method 551.1, Revision 1.0, "Determination of Chlorination Disinfection Byproducts, Chlorinated Solvents, and Halogenated Pesticides/Herbicides in Drinking Water by Liquid-Liquid Extraction and Gas Chromatography with Electron-Capture Detection" in Methods for the Determination of Organic Compounds in Drinking Water—Supplement III, EPA/600/R–95–131. (Available at http://www.nemi.gov.)

USEPA. 1995d. EPA Method 504.1, Revision 1.1, "1,2-Dibromoethane (EDB), 1,2-Dibromo-3-Chloro-Propane (DBCP), and 1,2,3-Trichloropropane (123TCP) in Water by Microextraction and Gas Chromatography" in Methods for the Determination of Organic Compounds in Drinking Water— Supplement III, EPA/600/R–95–131. (Available at http://www.nemi.gov.)

USEPA. 2007. Expedited Approval of Test Procedures for the Analysis of Contaminants Under the Safe Drinking Water Act; Analysis and Sampling Procedures. 72 FR 17902. April 10, 2007. USEPA. 2008. Expedited Approval of Alternative Test Procedures for the Analysis of Contaminants Under the Safe Drinking Water Act; Analysis and Sampling Procedures. 73 FR 31616. June 3, 2008.

USEPA. 2009. EPA Method 524.3 Version 1.0. Measurement of Purgeable Organic Compounds in Water by Capillary Column Gas Chromatography/ Mass Spectrometry, EPA 815–B–09–009. June 2009. (Available at http://epa.gov/ safewater/methods/

analyticalmethods_ogwdw.html.) Wendelken, S., 2009a. Memo to the

record describing basis for expedited approval of Mitchell Method M5271. May 29, 2009.

Wendelken, S., 2009b. Memo to the record describing basis for expedited approval of Mitchell Method M5331. May 29, 2009.

Wendelken, S., 2009c. Memo to the record describing ATP evaluation of

Thermo Scientific/Orion Method AQ4500, Revision 1.0 and basis for expedited approval. May 29, 2009.

Wendelken, S., 2009d. Memo to the record describing basis for expedited approval of Systea Easy (1–Reagent) Nitrate Method. May 29, 2009.

Wendelken, S., 2009e. Memo to the record describing ATP evaluation of Method ME355.01 and basis for expedited approval. May 29, 2009.

Zaffiro, A.D, Prakash, B. and Zimmerman, M., 2009. EPA Method 524.3 Research Summary, Shaw Environmental, Cincinnati OH. June 2009.

List of Subjects in 40 CFR Part 141

Environmental protection, Chemicals, Indians—lands, Intergovernmental relations, Radiation protection, Reporting and recordkeeping requirements, Water supply. Dated: July 9, 2009.

Michael H. Shapiro, Acting Assistant Administrator, Office of Water.

■ For the reasons stated in the preamble, 40 CFR part 141 is amended as follows:

PART 141—NATIONAL PRIMARY DRINKING WATER REGULATIONS

■ 1. The authority citation for part 141 continues to read as follows:

Authority: 42 U.S.C. 300f, 300g-1, 300j-4, and 300j-9.

■ 2. Subpart C is amended by revising Appendix A to read as follows:

Appendix A to Subpart C of Part 141— Alternative Testing Methods Approved for Analyses Under the Safe Drinking Water Act

Only the editions stated in the following table are approved.

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 141.21(f)(3)

Organism	Methodology	SM 21st edition ¹
Total Coliforms	Total Coliform Fermentation Technique Total Coliform Membrane Filter Technique Presence-Absence (P–A) Coliform Test ONPG–MUG Test	9221 A, B 9222 A, B, C 9221 D 9223

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 141.23(k)(1)

Contaminant	Methodology	EPA method	SM 21st edition ¹	SM online ³	ASTM ⁴	Other
Alkalinity Antimony	Titrimetric Atomic Absorption; Furnace Axially viewed inductively cou- pled plasma-atomic emission	200.5, Revision 4.2 ² .	2320 B 3113 B			
Arsenic	spectrometry (AVICP-AES). Atomic Absorption; Furnace Hydride Atomic Absorption Axially viewed inductively cou-	200.5, Revision	3113 B 3114 B			
Barium	pled plasma-atomic emission spectrometry (AVICP–AES). Inductively Coupled Plasma Atomic Absorption; Direct Atomic Absorption; Furnace	4.2.	3120 B 3111 D 3113 B			
Beryllium	Axially viewed inductively cou- pled plasma-atomic emission spectrometry (AVICP-AES). Inductively Coupled Plasma Atomic Absorption; Furnace Axially viewed inductively cou-	200.5, Revision 4.2. 200.5, Revision	3120 B 3113 B			
Cadmium	pled plasma-atomic emission spectrometry (AVICP–AES). Atomic Absorption; Furnace Axially viewed inductively cou- pled plasma-atomic emission	4.2. 200.5, Revision 4.2.	3113 B			
Calcium	spectrometry (AVICP–AES). EDTA titrimetric Atomic Absorption; Direct Aspi- ration.		3500–Ca B 3111 B			
	Inductively Coupled Plasma Axially viewed inductively cou- pled plasma-atomic emission spectrometry (AVICP-AES).	200.5, Revision 4.2.	3120 B			
Chromium	Inductively Coupled Plasma Atomic Absorption; Furnace		3120 B 3113 B			

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Contaminant	Methodology	EPA method	SM 21st edition ¹	SM online ³	ASTM ⁴	Other
	Axially viewed inductively cou- pled plasma-atomic emission spectrometry (AVICP–AES).	200.5, Revision 4.2.				
Copper	Atomic Absorption; Furnace Atomic Absorption; Direct Aspi-		3113 B 3111 B			
	ration. Inductively Coupled Plasma Axially viewed inductively cou- pled plasma-atomic emission	200.5, Revision 4.2.	3120 B			
Conductivity	spectrometry (AVICP–AES). Conductance	7.2.	2510 B			
yanide	Manual Distillation followed by Spectrophotometric, Amenable.		4500–CN [–] G		D2036–06 A D2036–06 B	
	Spectrophotometric Manual Selective Electrode		4500–CN − E 4500–CN − F		D2036–06 A	
lu a viela	Gas Chromatography/Mass Spectrometry Headspace.		4140 0			ME355.01 ⁷
luoride	Ion Chromatography Manual Distillation; Colorimetric SPADNS.		4110 B 4500–F [–] B, D			
aad	Manual Electrode Automated Alizarin		4500–F [–] C 4500–F [–] E		D1179–04 B	
ead	Atomic Absorption; Furnace Axially viewed inductively cou- pled plasma-atomic emission spectrometry (AVICP–AES).	200.5, Revision 4.2.	3113 B			
lagnesium	Atomic Absorption Inductively Coupled Plasma Complexation Titrimetric Meth-		3111 B 3120 B 3500–Mg B			
	ods. Axially viewed inductively cou- pled plasma-atomic emission	200.5, Revision 4.2.				
lercury	spectrometry (AVICP-AES). Manual, Cold Vapor		3112 B			
ickel	Inductively Coupled Plasma Atomic Absorption; Direct Atomic Absorption; Furnace Axially viewed inductively cou- pled plasma-atomic emission	200.5, Revision 4.2.	3120 B 3111 B 3113 B			
itrate	spectrometry (AVICP–AES). Ion Chromatography Automated Cadmium Reduction Manual Cadmium Reduction Ion Selective Electrode Reduction/Colorimetric		4110 B 4500–NO ₃ ⁻ F 4500–NO ₃ ⁻ E 4500–NO ₃ ⁻ D			Systea Easy
itrite	Ion Chromatography Automated Cadmium Reduction Manual Cadmium Reduction Spectrophotometric		4110 B 4500–NO ₃ ⁻ F 4500–NO ₃ ⁻ E 4500–NO ₂ ⁻ B			(1-Reagent) ²
	Reduction/Colorimetric		1000 1102 2			Systea Easy (1-Reagent)
rthophosphate	Ion Chromatography Colorimetric, ascorbic acid, sin-		4110 B 4500–P E	4500–P E–99		
	gle reagent. Colorimetric, Automated, Ascor- bic Acid.		4500–P F	4500–P F–99		
H elenium	Electrometric Hydride-Atomic Absorption Atomic Absorption; Furnace Axially viewed inductively cou-	200.5, Revision	4500–H⁺ B 3114 B 3113 B			
ilica	pled plasma-atomic emission spectrometry (AVICP–AES). Colorimetric	4.2.			D859–05	
	Molybdosilicate Heteropoly blue Automated for Molybdate-reac- tive Silica.		4500–SiO ₂ C 4500–SiO ₂ D 4500–SiO ₂ E			

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 141.23(k)(1)-Continued

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 141.23(k)(1)-Continued

Contaminant	Methodology	EPA method	SM 21st edition ¹	SM online ³	ASTM ⁴	Other
Sodium	Axially viewed inductively cou- pled plasma-atomic emission spectrometry (AVICP-AES). Inductively Coupled Plasma Atomic Absorption; Direct Aspi- ration. Axially viewed inductively cou- pled plasma-atomic emission spectrometry (AVICP-AES).	200.5, Revision	3120 B 3111 B			
Temperature	Thermometric		2550			

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 141.24(e)(1)

Contaminant	Methodology	EPA method	SM 21st edition ¹	SM online ³
Benzene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3 ⁹		
Carbon tetrachloride	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Chlorobenzene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
1,2-Dichlorobenzene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
1,4-Dichlorobenzene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
1,2-Dichloroethane	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
cis-Dichloroethylene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Trans-Dichloroethylene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Dichloromethane	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
1,2-Dichloropropane	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Ethylbenzene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Styrene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Tetrachloroethylene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
1,1,1-Trichloroethane	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Trichloroethylene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Toluene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
1,2,4-Trichlorobenzene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
1,1-Dichloroethylene	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
1,1,2-Trichlorethane	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Vinyl chloride	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Xylenes (total)	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Carbofuran	High-performance liquid chromatography (HPLC) with post- column derivatization and fluorescence detection.		6610 B	6610 B–04
Dibromochloropropane (DBCP).	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Ethyl dibromide (EDB)	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		
Oxamyl	High-performance liquid chromatography (HPLC) with post- column derivatization and fluorescence detection.		6610 B	6610 B–04
Total Trihalomethanes	Purge & Trap/Gas Chromatography/Mass Spectrometry	524.3		

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 141.25(a)

Contaminant	Methodology	SM 21st edition ¹	ASTM ⁴
Naturally Occurring:			
Gross alpha and beta	Evaporation	7110 B	
Gross alpha	Coprecipitation	7110 C	
Radium 226	Radon emanation	7500–Ra C	
	Radiochemical	7500–Ra B	
Radium 228	Radiochemical	7500–Ra D	
Uranium	Radiochemical	7500–U B	
	ICP-MS		D5673-05
	Alpha spectrometry	7500–U C	
Man-Made:			
Radioactive Cesium	Radiochemical	7500–Cs B	
	Gamma Ray Spectrometry	7120	
Radioactive Iodine	Radiochemical	7500–I B	
		7500–I C	
		7500–I D	
	Gamma Ray Spectrometry	7120	
Radioactive Strontium 89, 90	Radiochemical	7500–Sr B	
	Liquid Scintillation	7500–3H B	

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 141.25(a)—Continued

Contaminant	Methodology	SM 21st edition ¹	ASTM ⁴
Gamma Emitters	Gamma Ray Spectrometry	7120 7500–Cs B 7500–I B	

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 141.74(a)(1)

Organism	Methodology	SM 21st edition ¹	Other
Total Coliform	Total Coliform Fermentation Technique Total Coliform Membrane Filter Technique ONPG-MUG Test	9221 A, B, C 9222 A, B, C 9223	
Fecal Coliforms	Fecal Coliform Procedure Fecal Coliform Filter Procedure	9221 E 9222 D	
Heterotrophic bacteria Turbidity	Pour Plate Method Nephelometric Method Laser Nephelometry (on-line)	9215 B 2130 B	Mitchell M5271 ¹⁰
	LED Nephelometry (on-line) LED Nephelometry (portable)		Mitchell M5331 ¹¹ Orion AQ4500 ¹²

ALTERNATIVE TESTING METHODS FOR DISINFECTANT RESIDUALS LISTED AT 40 CFR 141.74(a)(2)

Residual	Methodology	SM 21st edition ¹
Free Chlorine	Amperometric Titration	4500–CI D
	DPD Ferrous Titrimetric	4500–CI F
	DPD Colorimetric	4500–CI G
	Syringaldazine (FACTS)	4500–CI H
Total Chlorine	Amperometric Titration	4500–CI D
	Amperometric Titration (Low level measurement)	4500–CI E
	DPD Ferrous Titrimetric	4500–CI F
	DPD Colorimetric	4500–CI G
	Iodometric Electrode	4500–CI I
Chlorine Dioxide	Amperometric Titration	4500–CIO ₂ C
	Amperometric Titration	4500–CIO ₂ E
Ozone	Indigo Method	4500–O ₃ B

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 141.131(b)(1)

Contaminant	Methodology	EPA method	SM 21st edition ¹
TTHM HAA5 Chlorite—daily monitoring as prescribed in 40 CFR 141.132(b)(2)(i)(A).	P&T/GC/MS LLE (diazomethane)/GC/ECD Amperometric Titration	524.3 ⁹	6251 B 4500–CIO ₂ E

ALTERNATIVE TESTING METHODS FOR DISINFECTANT RESIDUALS LISTED AT 40 CFR 141.131(c)(1)

Residual	Methodology	SM 21st edition ¹
Free Chlorine	Amperometric Titration	4500–CI D
	DPD Ferrous Titrimetric	4500–CI F
	DPD Colorimetric	4500–CI G
	Syringaldazine (FACTS)	4500–CI H
Combined Chlorine	Amperometric Titration	
	DPD Ferrous Titrimetric	4500–CI F
	DPD Colorimetric	4500–CI G
Total Chlorine	Amperometric Titration	4500–CI D
	Low level Amperometric Titration	
	DPD Ferrous Titrimetric	4500–CI F
	DPD Colorimetric	4500–CI G
	Iodometric Electrode	4500–CI I
Chlorine Dioxide	Amperometric Method II	4500–CIO ₂ E

ALTERNATIVE TESTING METHODS FOR DISINFECTANT RESIDUALS LISTED AT 40 CFR 141.131(c)(2), IF APPROVED BY THE STATE

Residual	Residual Methodology	
Free Chlorine	Test Strips	Method D99–003 ⁵

ALTERNATIVE TESTING METHODS FOR PARAMETERS LISTED AT 40 CFR 141.131(d)

Parameter	Methodology	SM 21st edition ¹
Total Organic Carbon (TOC)	High Temperature Combustion Persulfate-Ultraviolet or Heated Persulfate Oxidation Wet Oxidation	5310 B 5310 C 5310 D
Specific Ultraviolet Absorbance (SUVA) Dissolved Organic Carbon (DOC)	Calculation using DOC and UV ₂₅₄ data High Temperature Combustion	
Ultraviolet absorption at 254 nm (UV_{254}) .	Spectrophotometry	5910 B

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 141.402(c)(2)

Organism	Methodology	SM 20th edition ⁶	SM 21st edition ¹	SM online ³
<i>E. coli</i>	Colilert Colisure Colilert-18 Multiple-Tube Technique	9223 B	9223 B 9223 B 9223 B	9223 B-97 9223 B-97 9223 B-97 9230 B-04

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 141.704(b)

Organism	Methodology	SM 20th edition ⁶
E. coli	Membrane Filtration, Two Step	9222 D/9222 G

ALTERNATIVE TESTING METHODS FOR CONTAMINANTS LISTED AT 40 CFR 143.4(b)

Contaminant	Methodology	EPA method	ASTM 4	SM 21st edition ¹	SM online ³
Aluminum	Axially viewed inductively coupled plasma-atomic	200.5, Revision			
	emission spectrometry (AVICP–AES).	4.2 ² .			
	Atomic Absorption; Direct			3111 D	
	Atomic Absorption; Furnace			3113 B	
Oblavida	Inductively Coupled Plasma		D 510 04 D	3120 B	
Chloride	Silver Nitrate Titration Ion Chromatography		D 512–04 B	4500–CI [–] B 4110 B	
	Potentiometric Titration			4500–Cl ⁻ D	
Color				2120 B	
Foaming Agents				5540 C	
Iron	Axially viewed inductively coupled plasma-atomic	200.5, Revision			
	emission spectrometry (AVICP-AES).	4.2.			
	Atomic Absorption; Direct			3111 B	
	Atomic Absorption; Furnace			3113 B	
	Inductively Coupled Plasma			3120 B	
Manganese					
	emission spectrometry (AVICP-AES).	4.2.			
	Atomic Absorption; Direct			3111 B	
	Atomic Absorption; Furnace			3113 B	
Oder	Inductively Coupled Plasma			3120 B	
Odor Silver	Threshold Odor Test Axially viewed inductively coupled plasma-atomic	200.5, Revision		2150 B	
Silver	emission spectrometry (AVICP-AES).	4.2.			
	Atomic Absorption; Direct	4.2.		3111 B	
	Atomic Absorption; Furnace			3113 B	
	Inductively Coupled Plasma			3120 B	
Sulfate	Ion Chromatography			4110 B	
	Gravimetric with ignition of residue			4500–SO ₄ -2 C	4500-SO ₄ -2 C-9
	Gravimetric with drying of residue			4500–SO ₄ -2 D	4500-SO ₄ -2 D-9
	Turbidimetric method				4500-SO ₄ -2 E-9

ALTERNATIVE	TESTING METHOD	S FOR CONTAMINANTS	LISTED AT 40 CFF	143.4(b)—Continued
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Contaminant	Methodology	EPA method	ASTM ⁴	SM 21st edition ¹	SM online ³
Total Dissolved Solids. Zinc	 Automated methylthymol blue method Total Dissolved Solids Dried at 180 deg C Axially viewed inductively coupled plasma-atomic emission spectrometry (AVICP–AES). Atomic Absorption; Direct Aspiration Inductively Coupled Plasma 	200.5, Revision 4.2.		4500–SO ₄ ⁻² F 2540 C 3111 B 3120 B	4500–SO ₄ ⁻² F–97

¹ Standard Methods for the Examination of Water and Wastewater, 21st edition (2005). Available from American Public Health Association, 800 I Street, NW., Washington, DC 20001–3710.

² EPA Method 200.5, Revision 4.2. "Determination of Trace Elements in Drinking Water by Axially Viewed Inductively Coupled Plasma-Atomic Emission Spectrometry." 2003. EPA/600/R– 06/115. (Available at http://www.epa.gov/ nerlcwww/ordmeth.htm.)

³ Standard Methods Online are available at *http://www.standardmethods.org.* The year in which each method was approved by the Standard Methods Committee is designated by the last two digits in the method number. The methods listed are the only online versions that may be used.

⁴ Available from ASTM International, 100 Barr Harbor Drive, West Conshohocken, PA 19428–2959 or *http://astm.org*. The methods listed are the only alternative versions that may be used.

⁵Method D99–003, Revision 3.0. "Free Chlorine Species (HOCl⁻ and OCl⁻) by Test Strip," November 21, 2003. Available from Industrial Test Systems, Inc., 1875 Langston St., Rock Hill, SC 29730.

⁶ Standard Methods for the Examination of Water and Wastewater, 20th edition (1998). Available from American Public Health Association, 800 I Street, NW., Washington, DC 20001–3710.

⁷ Method ME355.01, Revision 1.0. "Determination of Cyanide in Drinking Water by GC/MS Headspace," May 26, 2009. Available at *http://www.nemi.gov* or from James Eaton, H & E Testing Laboratory, 221 State Street, Augusta, ME 04333. (207) 287– 2727.

⁸ Systea Easy (1-Reagent). "Systea Easy (1-Reagent) Nitrate Method," February 4, 2009. Available at *http://www.nemi.gov* or from Systea Scientific, LLC., 900 Jorie Blvd., Suite 35, Oak Brook, IL 60523.

⁹EPA Method 524.3, Version 1.0. "Measurement of Purgeable Organic Compounds in Water by Capillary Column Gas Chromatography/Mass Spectrometry," June 2009. EPA 815–B–09–009. Available at http://epa.gov/safewater/methods/ analyticalmethods ogwdw.html.

¹⁰ Mitchell Method M5271, Revision 1.1. "Determination of Turbidity by Laser Nephelometry," March 5, 2009. Available at *http://www.nemi.gov* or from Leck Mitchell, PhD, PE, 656 Independence Valley Dr., Grand Junction, CO 81507.

¹¹ Mitchell Method M5331, Revision 1.1. "Determination of Turbidity by LED Nephelometry," March 5, 2009. Available at *http://www.nemi.gov* or from Leck Mitchell, PhD, PE, 656 Independence Valley Dr., Grand Junction, CO 81507.

¹² Orion Method AQ4500, Revision 1.0.
"Determination of Turbidity by LED Nephelometry," May 8, 2009. Available at *http://www.nemi.gov* or from Thermo Scientific, 166 Cummings Center, Beverly, MA 01915, *http://www.thermo.com*.
[FR Doc. E9–18361 Filed 7–31–09; 8:45 am]

BILLING CODE S

DEPARTMENT OF HOMELAND SECURITY

Federal Emergency Management Agency

44 CFR Part 64

[Docket ID FEMA-2008-0020; Internal Agency Docket No. FEMA-8085]

Suspension of Community Eligibility

AGENCY: Federal Emergency Management Agency, DHS. **ACTION:** Final rule.

SUMMARY: This rule identifies communities, where the sale of flood insurance has been authorized under the National Flood Insurance Program (NFIP), that are scheduled for suspension on the effective dates listed within this rule because of noncompliance with the floodplain management requirements of the program. If the Federal Emergency Management Agency (FEMA) receives documentation that the community has adopted the required floodplain management measures prior to the effective suspension date given in this rule, the suspension will not occur and a notice of this will be provided by publication in the Federal Register on a subsequent date.

DATES: *Effective Dates:* The effective date of each community's scheduled suspension is the third date ("Susp.") listed in the third column of the following tables.

FOR FURTHER INFORMATION CONTACT: If you want to determine whether a particular community was suspended

on the suspension date or for further information, contact David Stearrett, Mitigation Directorate, Federal Emergency Management Agency, 500 C Street, SW., Washington, DC 20472, (202) 646–2953.

SUPPLEMENTARY INFORMATION: The NFIP enables property owners to purchase flood insurance which is generally not otherwise available. In return, communities agree to adopt and administer local floodplain management aimed at protecting lives and new construction from future flooding. Section 1315 of the National Flood Insurance Act of 1968, as amended, 42 U.S.C. 4022, prohibits flood insurance coverage as authorized under the NFIP, 42 U.S.C. 4001 et seq.; unless an appropriate public body adopts adequate floodplain management measures with effective enforcement measures. The communities listed in this document no longer meet that statutory requirement for compliance with program regulations, 44 CFR part 59. Accordingly, the communities will be suspended on the effective date in the third column. As of that date, flood insurance will no longer be available in the community. However, some of these communities may adopt and submit the required documentation of legally enforceable floodplain management measures after this rule is published but prior to the actual suspension date. These communities will not be suspended and will continue their eligibility for the sale of insurance. A notice withdrawing the suspension of the communities will be published in the Federal Register.

In addition, FEMA has identified the Special Flood Hazard Areas (SFHAs) in these communities by publishing a Flood Insurance Rate Map (FIRM). The date of the FIRM, if one has been published, is indicated in the fourth column of the table. No direct Federal financial assistance (except assistance pursuant to the Robert T. Stafford Disaster Relief and Emergency Assistance Act not in connection with a flood) may legally be provided for construction or acquisition of buildings